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Any polyatomic ion with the suffix “ -ate ” uses the suffix “ -ic ” as an acid. So, HNO₃ will be nitric acid. When you have a polyatomic ion with one more oxygen than the “ -ate ” ion, then your acid will have the prefix “ per- ” and the suffix “ -ic. ” For example, the chlorate ion is ClO₃⁻. Therefore, HClO₄ is called perchloric acid.

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