

## Modern Chemistry Review Answers

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Modern Chemistry Review Answers

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope. [http://mchsapchemistry.com/uploads/7/7/1/5/7715124/ch\\_3\\_study\\_guide\\_key.p...](http://mchsapchemistry.com/uploads/7/7/1/5/7715124/ch_3_study_guide_key.p...)

Modern Chemistry Chapter 5 Review Answers Section 3

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Modern Chemistry 89 States Of Matter CHAPTER 10 REVIEW States of Matter SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Indicate whether each of the following is a physical or chemical property of water.

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### Chapter 7 Section 3 Modern Chemistry Review Answers

Answers in Appendix E 1. A balloon ~ lled with helium gas has a volume of 500 mL at a pressure of 1 atm. ° e balloon is released and reaches an altitude of 6.5 km, where the pressure is 0.5 atm. If the temperature has remained the same, what volume does the gas occupy at this height?

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom ' s

### 5 The Periodic Law

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