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Write the formula for the ionic compound, using subscripts to indicate the number of each kind of ion used (from Columns 2 and 3). Follow the rules for writing ionic formulas listed in the background material.

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our example, the formula is  $AlCl_3$  Record the chemical formula for the  
...

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Read PDF Ionic Formula Writing Kit Answers and solutions on writing ionic equations. Example: Write the ionic equation for the word equation. Sodium chloride(aq) + silver nitrate(aq) → silver chloride(s) + sodium nitrate(aq) Solution: Step 1: Write the equation and balance it if necessary .  $\text{NaCl(aq)} + \text{AgNO}_3 \text{(aq)} \rightarrow \text{AgCl(s)} + \text{NaNO}_3 \text{(aq)}$  Step 2: Split the

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6. Write the formula for the ionic compound, using subscripts to indicate the number of each kind of ion used (from Columns 2 and 3). Follow the rules for writing ionic formulas listed in the background

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material. In our example, the formula is  $\text{AlCl}_3$ . Record the chemical formula for the ionic compound in Column 6 of the data table. 7.

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