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CHEMISTRY II

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Their Properties

Note

Properties of
Nonmetals

- Phosphorus, sulfur, and chlorine are among these other elements found in your body.

- These elements

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are classified as
nonmetals. 19.2

Nonmetals

- Nonmetals are elements that usually are gases or brittle solids at room temperature.

Chapter 19:
Elements and Their
Properties

Each element has
its own unique

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Elements And

Properties. Each

contains a different number of protons and neutrons, giving

it its own atomic

number and mass

number. The atomic

number of an

element is equal to

the number of

protons that

element contains.

The mass number

is the number of

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Elements And

Their Properties

protons plus the
number of neutrons
of that element.

Therefore, it is

possible to

determine the

number of neutrons

by subtracting the

atomic number from

the mass number.

Properties of

Elements | Biology

for Non-Majors I

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Elements And

Their Properties

character is used to define the chemical properties that

metallic elements

present. Generally,

metals tend to lose

electrons to form

cations. Nonmetals

tend to gain

electrons to form

anions. They also

have a high

oxidation potential

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Elements And

Therefore they are easily oxidized and are strong reducing agents.

Answers

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Atoms and the

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Elements And

Their Properties

Atom H, He, B, Be,
I, F, Br, Es Lr.
Physical Science.
Notes. Properties of
Atoms and the
Periodic Table.

Structure of the
Atom. Elements are
abbreviated in
scientific shorthand.

- Symbols on the periodic table are short or abbreviated ways to

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Elements And

write the name of
an element.

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Elements: The
Carbon Family.The
Nitrogen
Family.Group 16
Elements : The

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Properties of
elements are the
periodic function to
their atomic
numbers. The
periodicity in

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Properties is due to
repetition of similar
outer shell
electronic

Answers

configuration at a
certain regular
intervals. In modern
periodic table is
based on modern
periodic law in
which elements are
arranged in
increasing order of
their atomic

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Note Taking Worksheet Classification of Elements & Periodicity in Properties ...

Modern periodic law: This law is based on the assumption that the physical and chemical properties of elements are the periodic function of

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Characteristics:

The 9 vertical columns present were called group. 1 st to VIII th group +0 group of inert gases. Ramsay introduced the inert gases. The 7 horizontal series termed as periods.

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Elements: Notes for
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elements.

substances with
only one type of
atom. metal. an
element that is
malleable, polishes
to a luster,
conducts heat and
electricity well, and
tends to lose

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electrons in a
chemical reaction.
metalloid. an
element with
properties of both
metals and
nonmetals.
nonmetal.

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Chapter 3 •

Genesis of Periodic
Classification

Dobereiner ' s

Triads In 1829,

Dobereiner

arranged certain

elements with

similar properties in

groups of three in

such a way that the

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atomic mass of the
middle element was
nearly the same as
the average atomic
masses of the first
and the third
elements.

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Periodicity in
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properties note

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answers as you
such as. By
searching the title,
publisher, or
authors of guide
you essentially
want, you can
discover them
rapidly. In the
house, workplace,
or perhaps in your
method can be
every best place

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Answers

Element families
are indicated by
numbers located at
the top of the
periodic table. Todd
Helmenstine. An

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Elements And

Their Properties
element family is a set of elements sharing common properties.

Elements are classified into families because the three main categories of elements (metals, nonmetals, and semimetals) are very broad. The characteristics of

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Their Properties
the elements in
these families are
determined
primarily by the
number of electrons
in the outer energy
shell.

Element Families of
the Periodic Table -
ThoughtCo
Mendeleev
predicted the
existence and

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Germanium (Ge)
and Gallium (Ga).

When they were discovered, their properties were found to closely match those predicted by Mendeleev; 1913: Henry Moseley discovers concept of atomic numbers. Observed that

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Xrays were
different for each
element.

Answers

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Answers

arrange the

elements originally

by atomic weight in

a scheme that

would show

similarity among

groups"what are

the answers for

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what are the

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Answers for chapter
19 note taking
worksheet on the
elements and their
properties' 'import
spring config file
based on property
in properties

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Presently, there are
118 elements

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Known to us. Out of
these 118 chemical
elements, some
elements are man-
made. When there
were only 31
elements it was
relatively easy to
study the
properties of these
chemical elements
individually.

Periodic

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Meyer presented
the classification of
elements in the
form of a curve

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between atomic
volume and atomic
masses and state
that the properties
of the elements are
the periodic
functions of their
atomic volumes.

[Here, atomic
volume = molecular
mass / density He
concluded that the
elements with
similar properties

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