

Chapter 11 Solutions Thermodynamics

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Step-by-step solution: Step 1 of 2 Carnot cycle is totally reversible cycle that consists of two reversible isothermal and two reversible isentropic processes. It has a maximum efficiency for given temperature limits.

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NCERT Solution of Thermodynamics Chemistry Class 11. Solution - This question requires students to find the enthalpy of combustion of methane, graphite and hydrogen at 298k, -890.3 kJ mol⁻¹, -393.5 kJ mol⁻¹ and -285.8 kJ mol⁻¹. Here enthalpy of CH₄ G is needed to be determined. Thermodynamics Class 11 NCERT.

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